

Mathematical Cell Biology Graduate Summer Course
University of British Columbia, May 1-31, 2012
Leah Edelstein-Keshet

**Microtubules,
polymer size distribution
and other balance eqn models**



www.math.ubc.ca/~keshet/MCB2012/

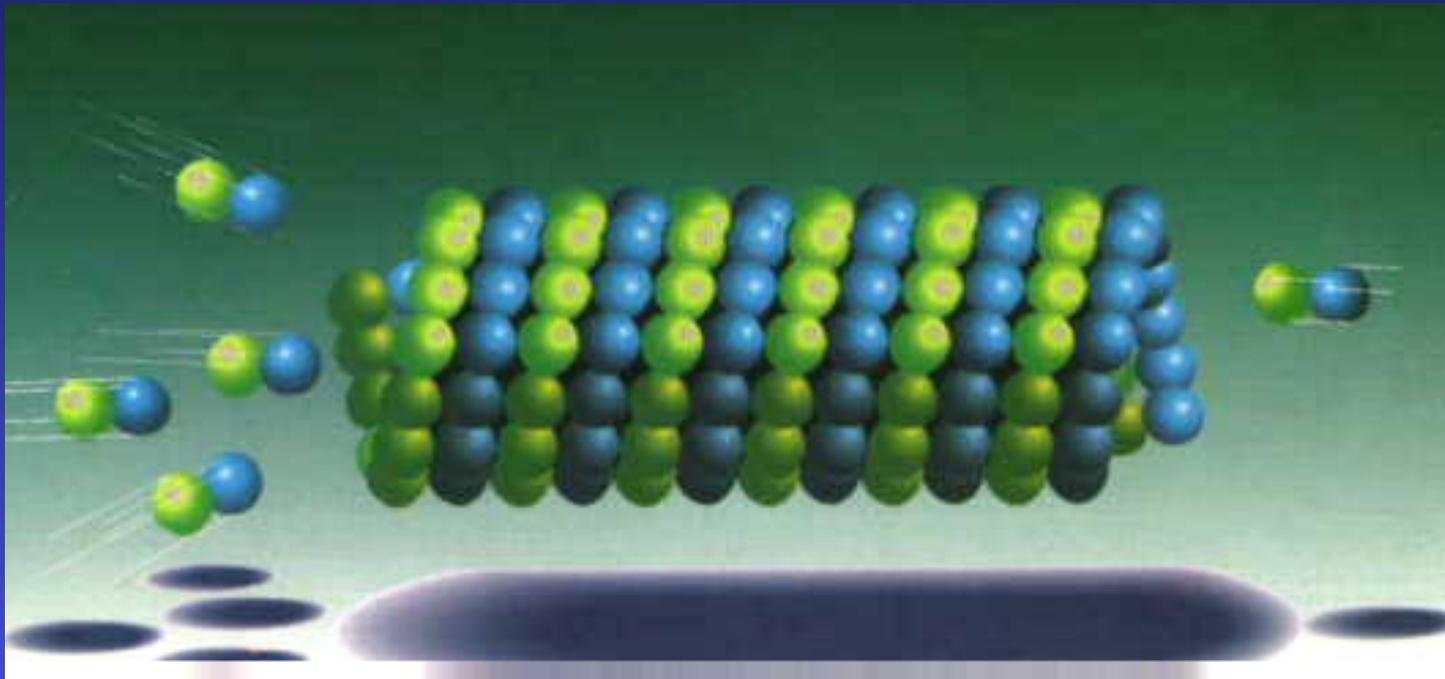


Microtubule (MT) dynamics

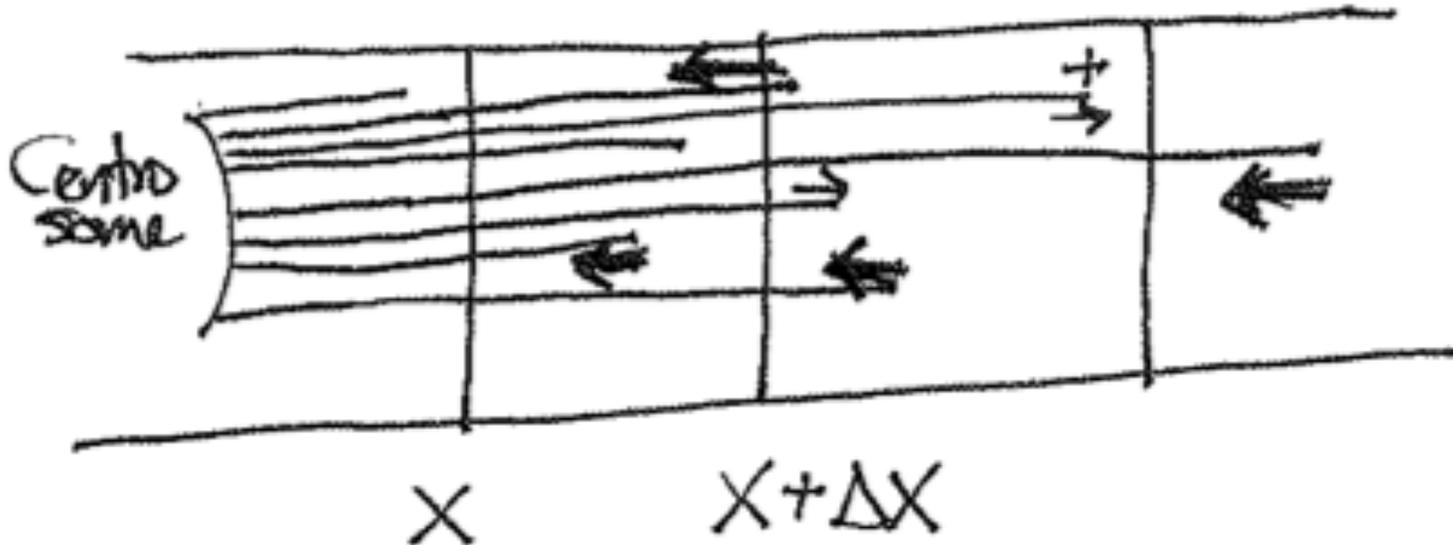
Actin filament:



Microtubule:



Growing and shrinking MT



Some Movies.....

<http://www.youtube.com/watch?v=9iXoXzgmEXw>

http://www.youtube.com/watch?v=PCI_GUHJJaY

MT dynamics

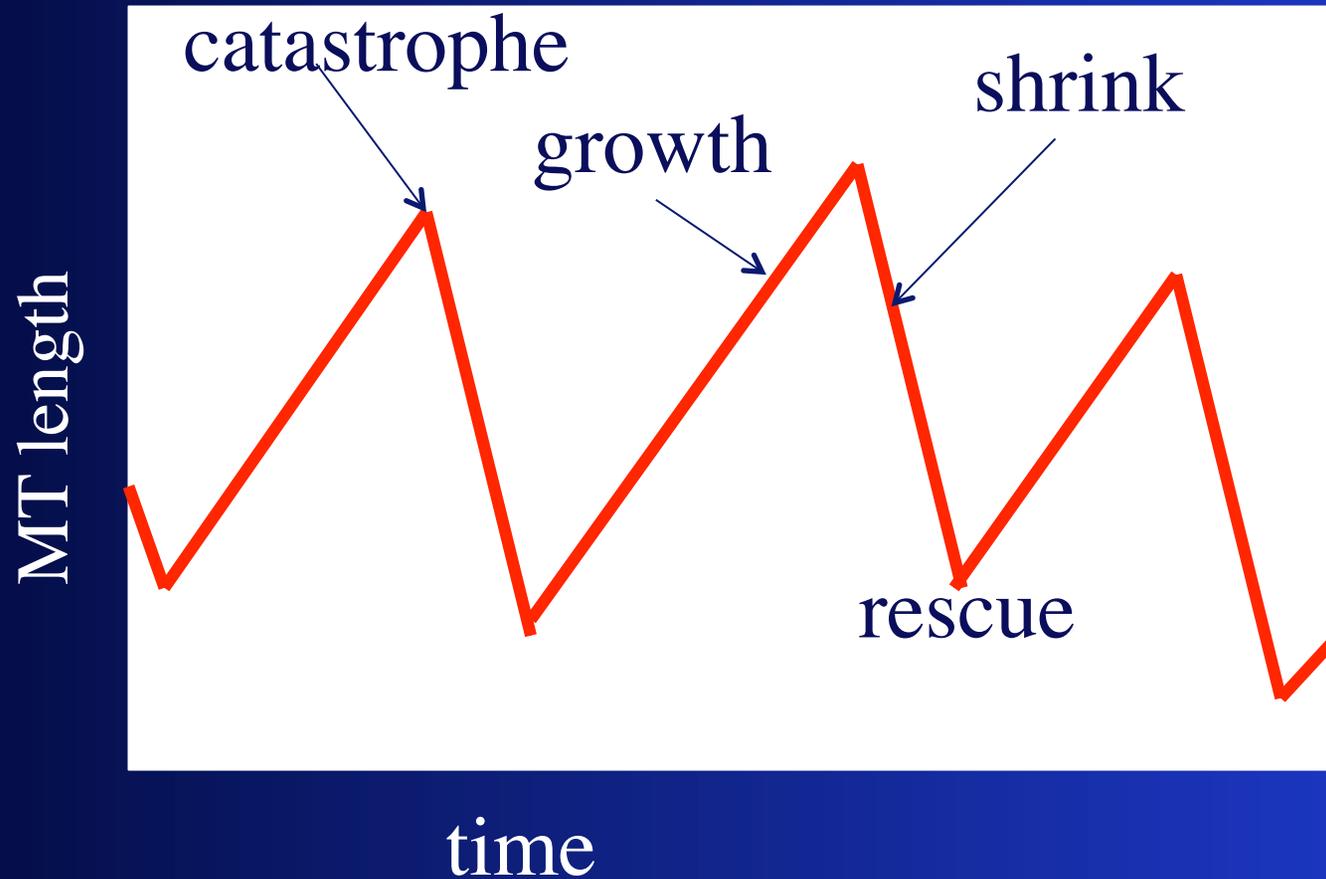
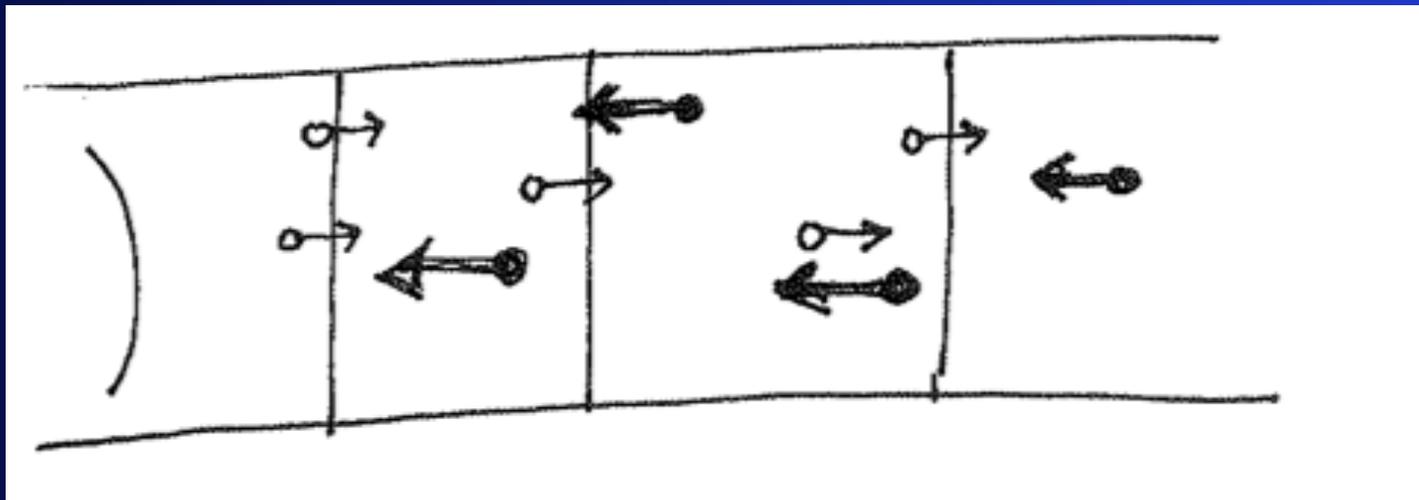
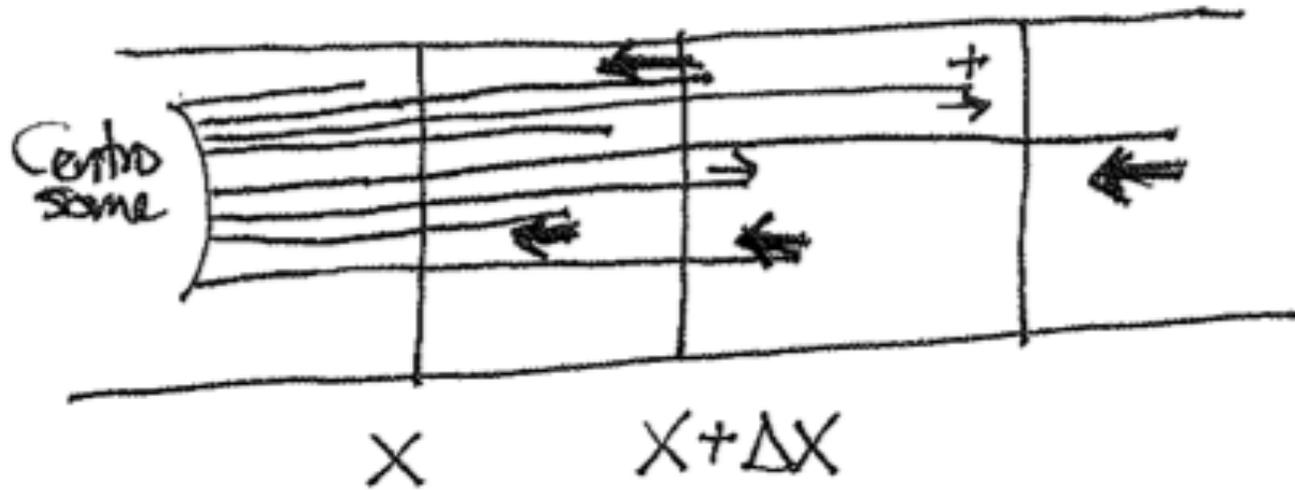


Figure adapted from

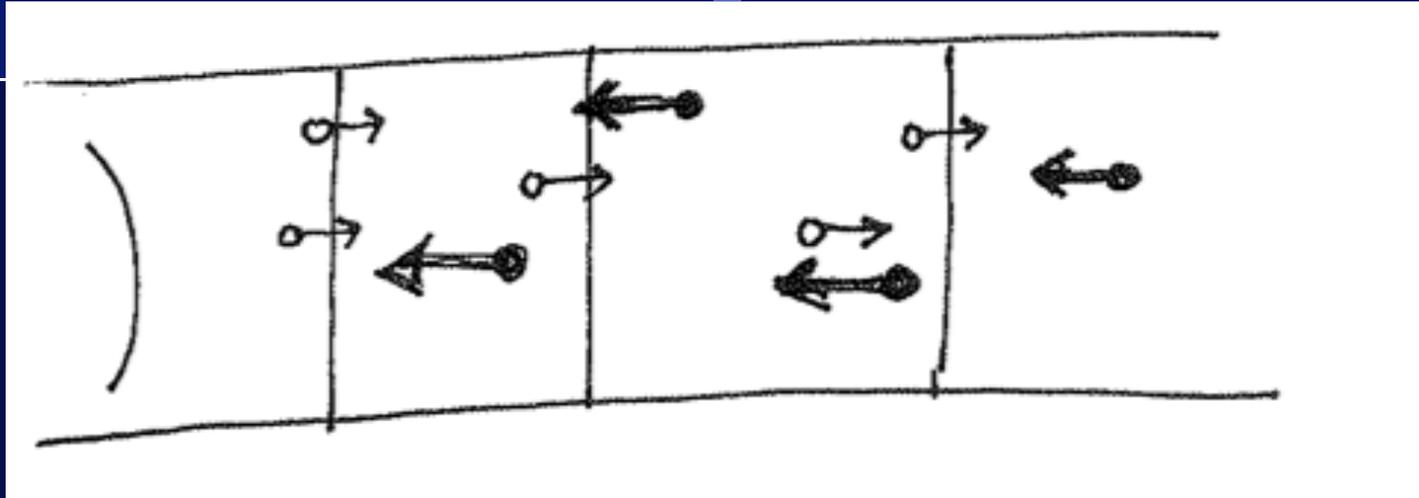
Janulevicius

Biophysical Journal Volume 90 February 2006 788-798

Growing and shrinking tips



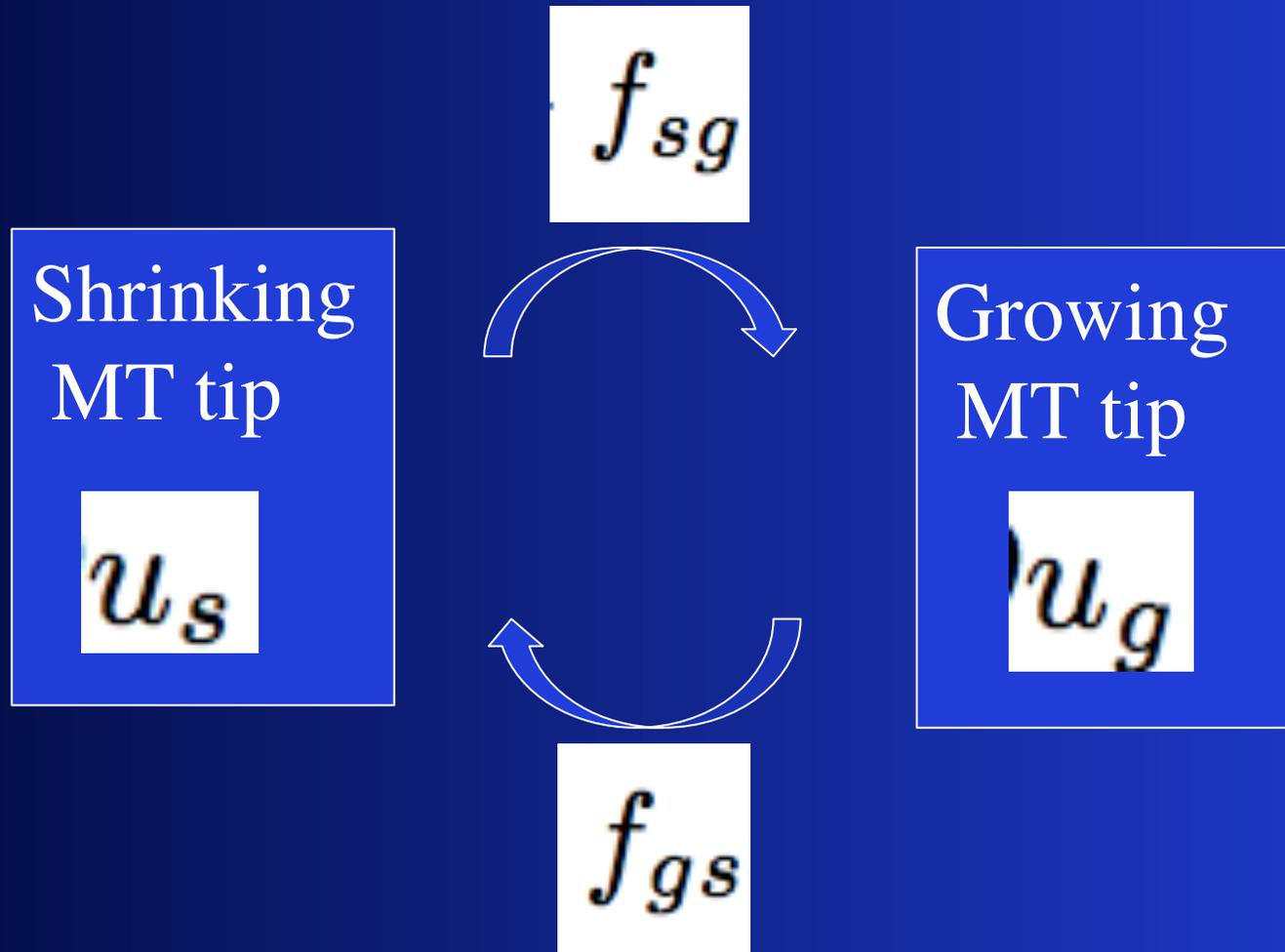
Balance equations



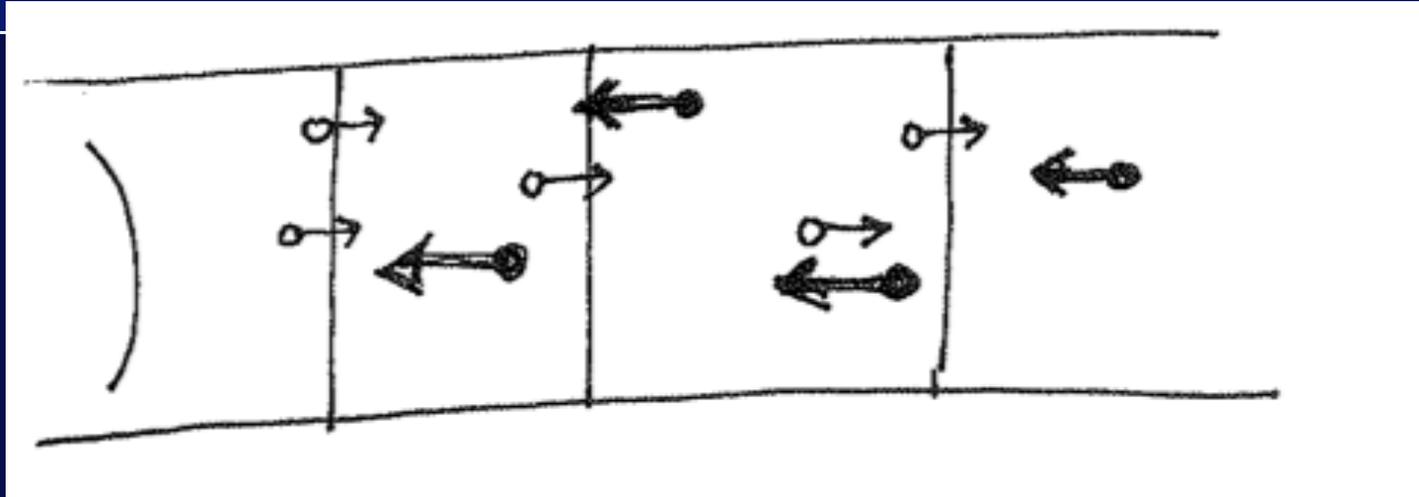
M. Dogterom and S. Leibler. Physical aspects of the growth and regulation of microtubule structures. *Phys. Rev Lett.*, 70:1347–1350, 1993.

M Dogterom, AC Maggs, and S Leibler. Diffusion and Formation of Microtubule Asters: Physical Processes Versus Biochemical Regulation. *PNAS*, 92(15):6683–6688, 1995.

Catastrophe and rescue



Tip fluxes:



$$J_g = u_g v_g, \quad J_s = -u_s v_s$$

Balance equations

$$\frac{\partial u_g}{\partial t} = -v_g \frac{\partial u_g}{\partial x} - f_{gs}u_g + f_{sg}u_s,$$

$$\frac{\partial u_s}{\partial t} = v_s \frac{\partial u_s}{\partial x} + f_{gs}u_g - f_{sg}u_s,$$

Spatial
terms

Exchange kinetics

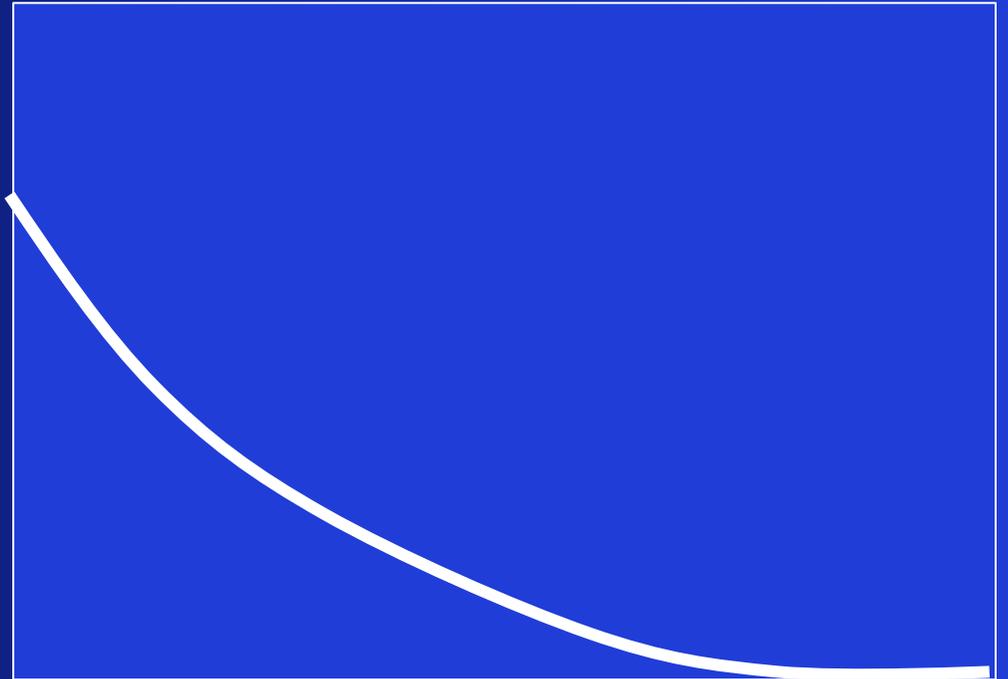
Steady state equations:

$$\frac{du_g}{dx} = \frac{1}{v_g} (-f_{gs}u_g + f_{sg}u_s),$$

$$\frac{du_s}{dx} = \frac{1}{v_s} (-f_{gs}u_g + f_{sg}u_s),$$

Behaviour

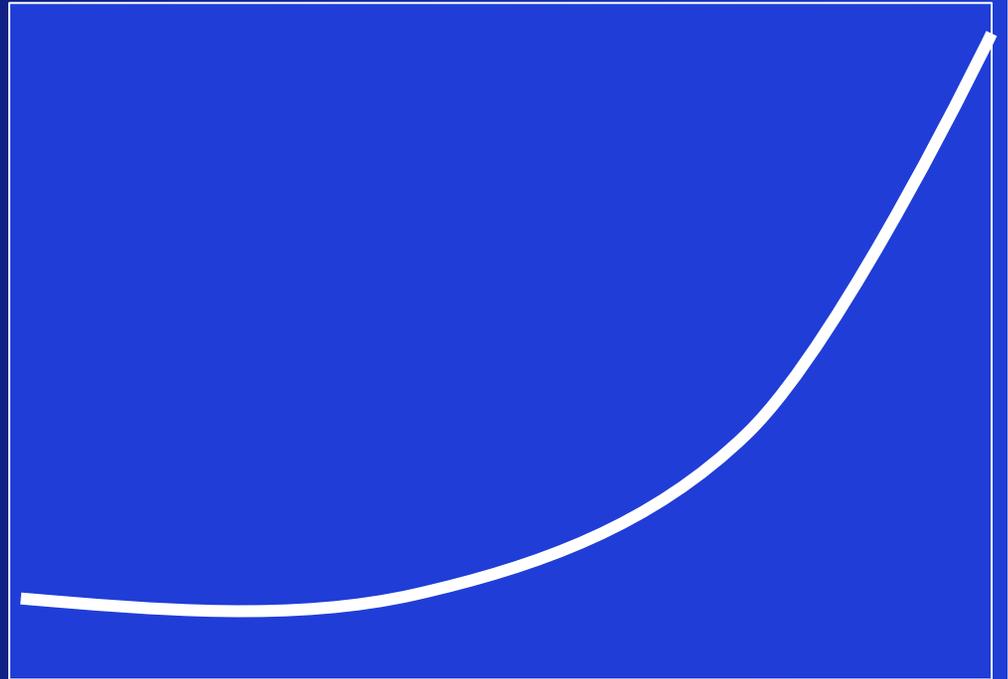
$$\frac{f_{sg}}{v_s} < \frac{f_{gs}}{v_g}$$



Distance from centrosome, x

Behaviour

$$\frac{f_{sg}}{v_s} > \frac{f_{gs}}{v_g}$$



Distance from centrosome, x



Experimental work (Komarova)

Life cycle of MTs: persistent growth in the cell interior, asymmetric transition frequencies and effects of the cell boundary

Yulia A. Komarova^{1,2,*}, Ivan A. Vorobjev² and Gary G. Borisy¹

Accepted 20 June 2002
Journal of Cell Science 115, 3527-3539

A decorative graphic in the top-left corner of the slide. It features a glowing blue sphere with a bright white center, positioned at the intersection of a vertical white line and a horizontal white line. The background is a gradient of blue, with a darker blue at the top and a lighter blue at the bottom.

Now back to the beginning

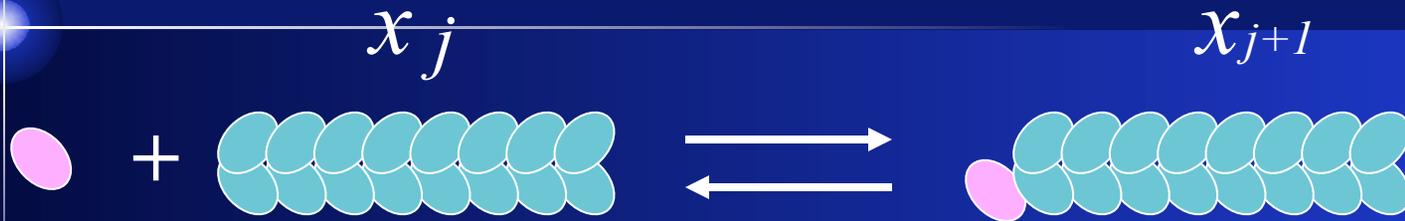


Polymer size distribution

Filament size distribution

It is very common in math-biology to consider size classes and formulate equations for the dynamics of size distributions (or age distributions, or distribution of some similar property).

Number of filaments of length j :



$$\frac{dx_j(t)}{dt} = \underbrace{k^+ a x_{j-1}}_{\text{Growth of shorter filament}} - \underbrace{(k^- + a k^+) x_j}_{\text{Monomer loss or gain}} + \underbrace{k^- x_{j+1}}_{\text{Shrinking of longer filament}}$$

Growth
of shorter
filament

Monomer
loss or
gain

Shrinking
of longer
filament

Steady state size distribution for constant pool of monomer

$$\frac{dx_j(t)}{dt} = k^+ a x_{j-1} - (k^- + a k^+) x_j + k^- x_{j+1}$$

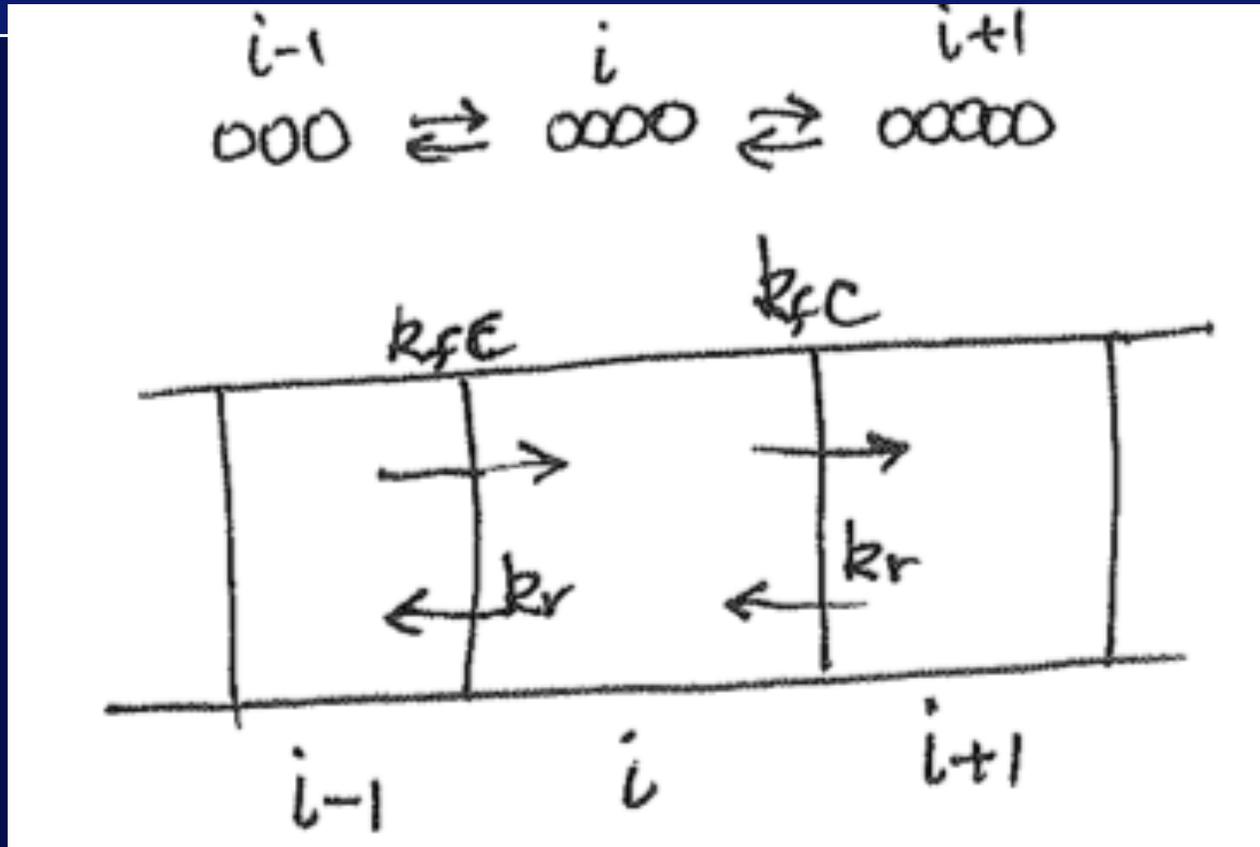
Find the steady state size distribution (assume that a , k^+ , k^- are constant.)

Express this in terms of $r = a k^+ / k^-$

To consider

- The solution will have some arbitrary constant(s). This means we need some additional constraint(s)..
- What do we use?
- (Consider the case that polymerization has to be started by cluster of n monomers)

Broader context

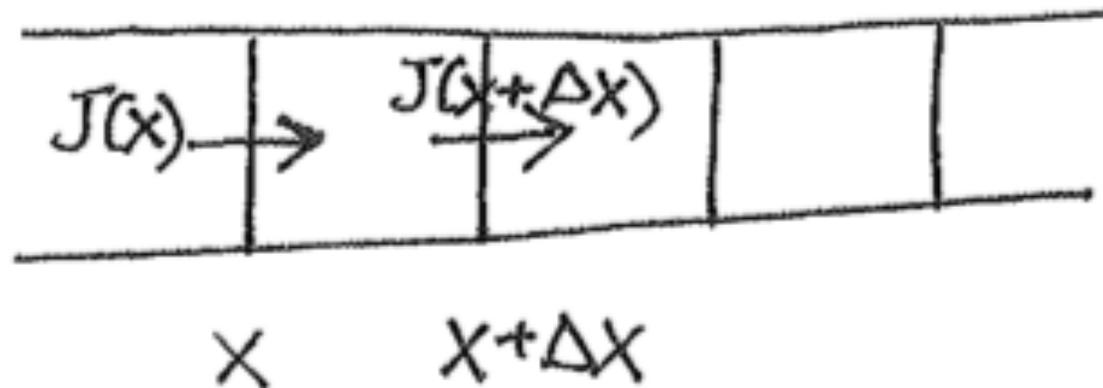


“Motion between size classes”

Keeping track of how many in class i

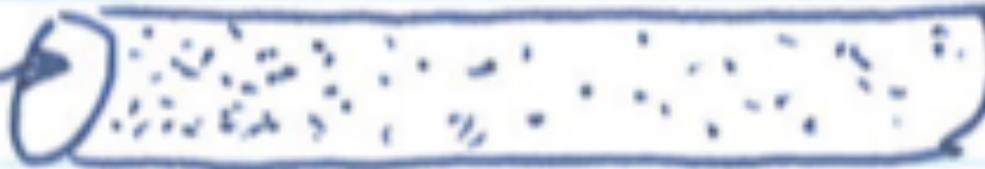
$$\begin{array}{l} \text{Rate of} \\ \text{change in} \\ \text{class } i \end{array} = \begin{array}{l} \text{Rate} \\ \text{entry} \\ \text{from} \\ \text{class } i-1 \end{array} + \begin{array}{l} \text{Rate} \\ \text{entry} \\ \text{from} \\ \text{class} \\ i+1 \end{array} + \begin{array}{l} \text{Rate loss} \\ \text{from class} \\ i \text{ to } i-1 \\ \text{and } i+1 \end{array}$$

Other balance equations



Continuous version

$$area = A$$



1D "tube" of length L

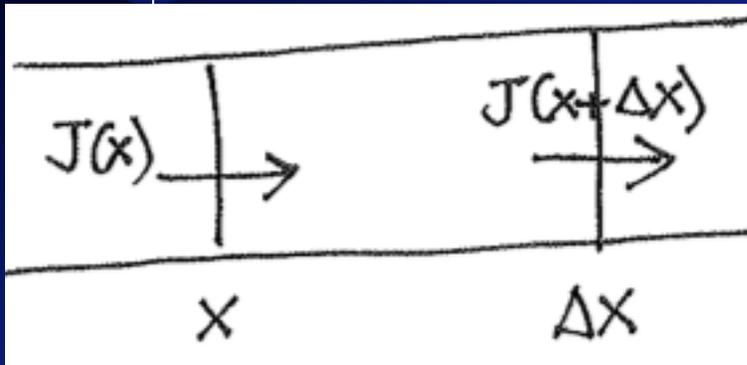
Flux

$\vec{J}(x, t) =$ ^{net} number of particles
crossing a unit area
at x and time t in
+ve x direction



Note: \vec{J} is a vector
whose magnitude
has units $(\text{area})^{-1}(\text{time})^{-1}$

Consider a small segment
between x and Δx



Concentration = $c(x, t)$,

Number = $c(x, t)A\Delta x$,

$$\frac{d(\text{Number})}{dt} = [\text{Rate in} - \text{rate out}] + \text{rate produced}$$

Balance equation

$$\frac{d(\text{Number})}{dt} = [\text{Rate in} - \text{rate out}] + \text{rate produced}$$

$$\frac{d(A\Delta xc)}{dt} = A [J(x) - J(x + \Delta x)] + (A\Delta x)\sigma$$

$$\frac{dc}{dt} = \frac{1}{\Delta x} [J(x) - J(x + \Delta x)] + \sigma$$

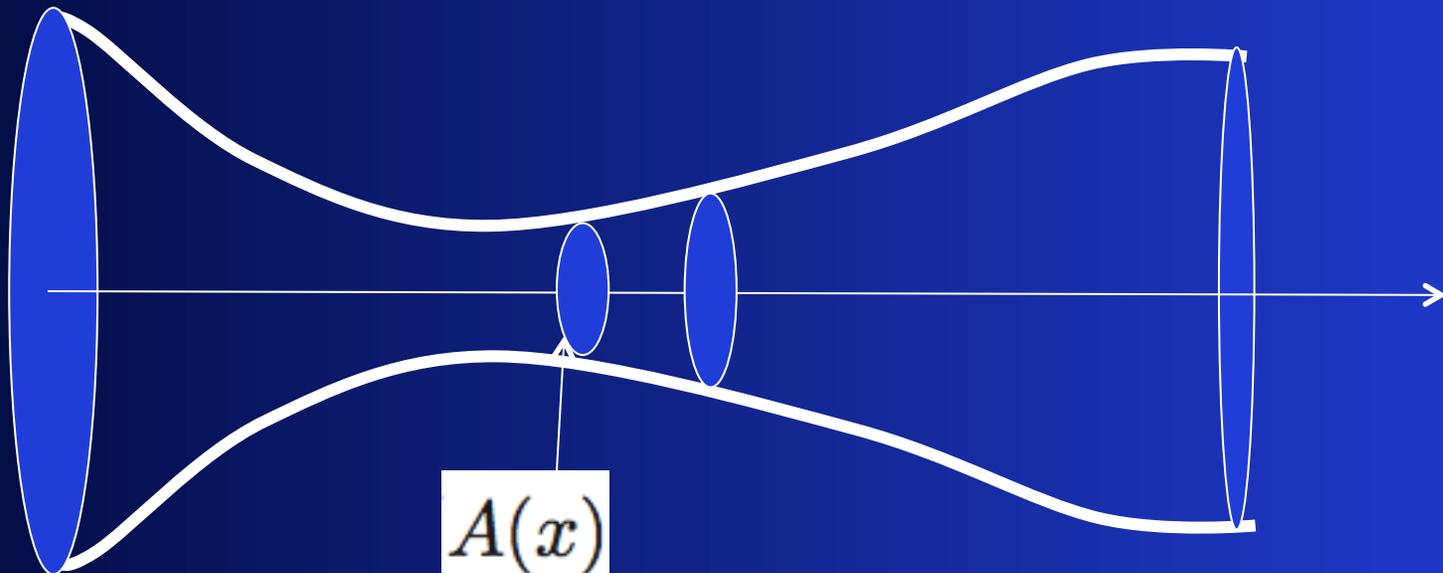
In the limit as $\Delta x \rightarrow 0..$

$$\frac{dc}{dt} = \frac{1}{\Delta x} [J(x) - J(x + \Delta x)] + \sigma$$

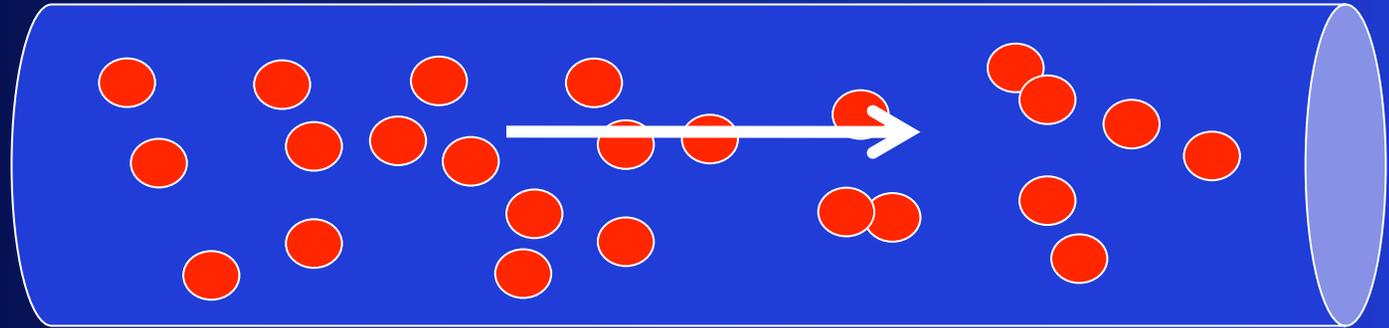
$$\frac{\partial c}{\partial t} = -\frac{\partial J}{\partial x} + \sigma$$

Nonconstant “tube” diameter

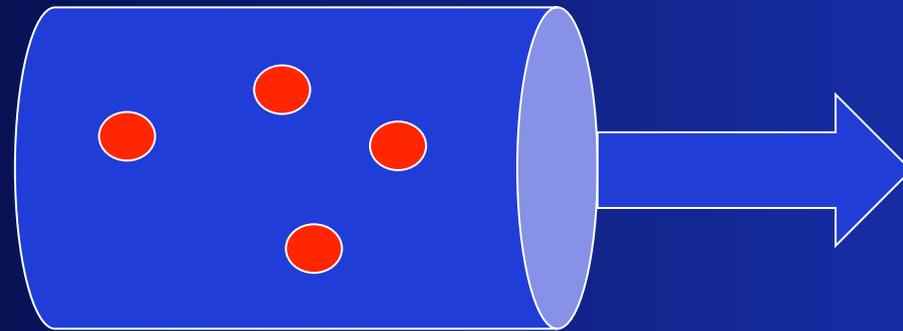
$$\frac{d(A(x)\Delta xc)}{dt} = [A(x)J(x) - A(x + \Delta x)J(x + \Delta x)] + (A(x)\Delta x)\sigma$$



Transport at velocity v

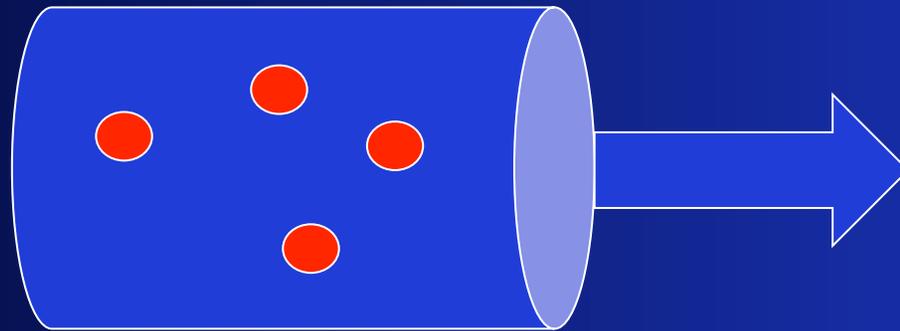


Transport at velocity v



What is
the flux?

Transport at velocity v

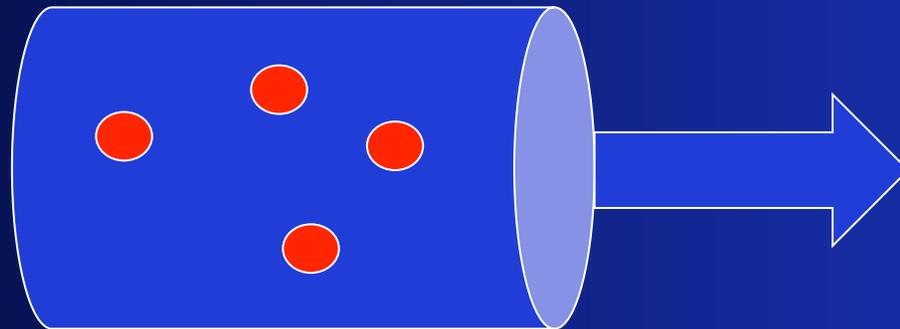


Volume $A v \Delta t$

Concentration c so $(c A v \Delta t)$ molecules
cross during time Δt

Flux $J = v c$

Convective flux (transport)



$$\text{Convective flux} = J_c = vc.$$

Transport equation

Convective flux = $J_c = vc$.

$$\frac{\partial c}{\partial t} = -\frac{\partial vc}{\partial x} + \sigma$$

Diffusion: Fick's Law

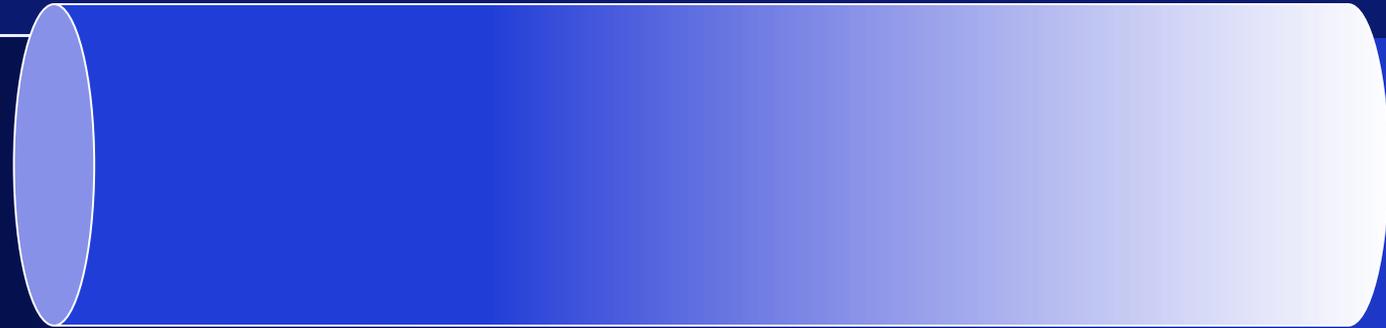
Flux proportional to concentration gradient



$$J = -D \frac{\partial c}{\partial x}$$

D = diffusion coefficient

Diffusion



$$\frac{\partial c}{\partial t} = -\frac{\partial}{\partial x} \left(-D \frac{\partial c}{\partial x} \right) + \sigma$$

$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2} + \sigma$$

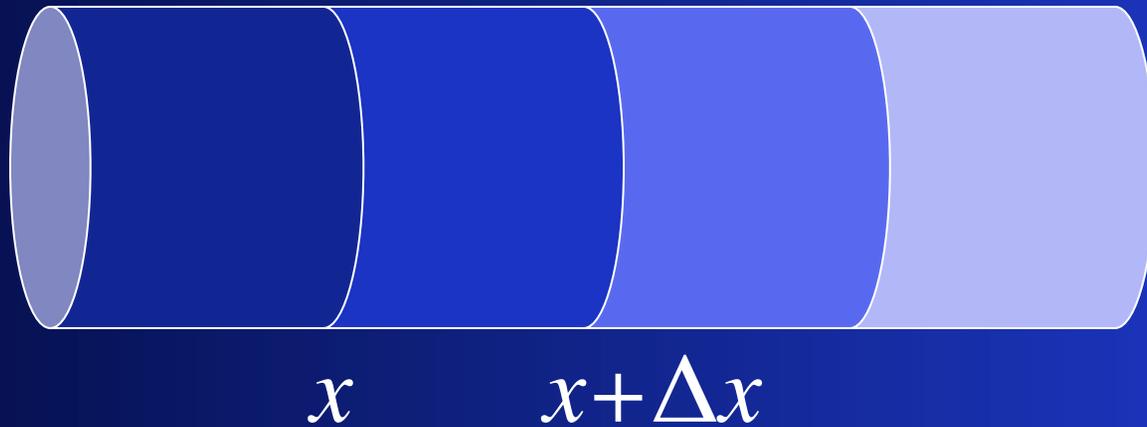
Boundaries

Typical Boundary Conditions

Q: How many are needed? A: This depends on the order of the highest ^{spatial} derivative.

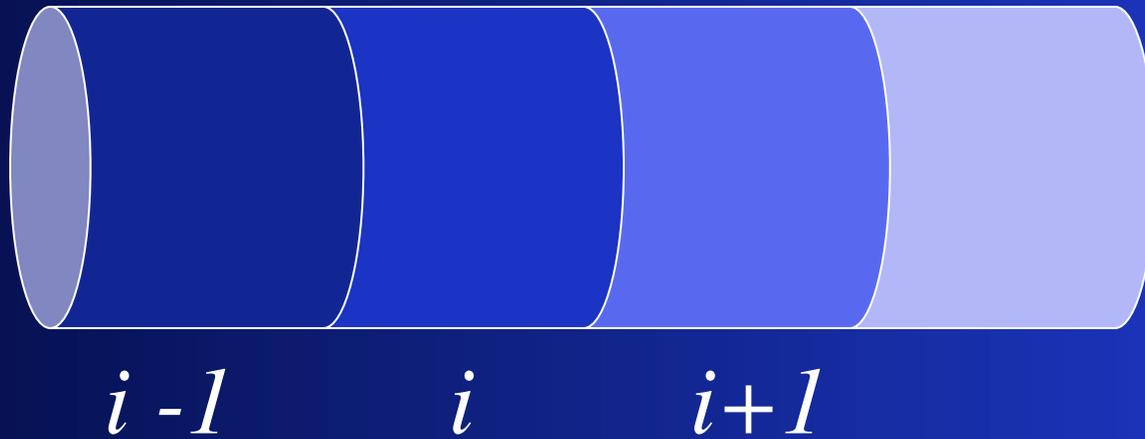
- Presence of diffusion term \Leftrightarrow 2nd derivative $\frac{\partial^2 c}{\partial x^2}$
 \Leftrightarrow two BC's needed, one for each domain end.

Discrete Diffusion Equation



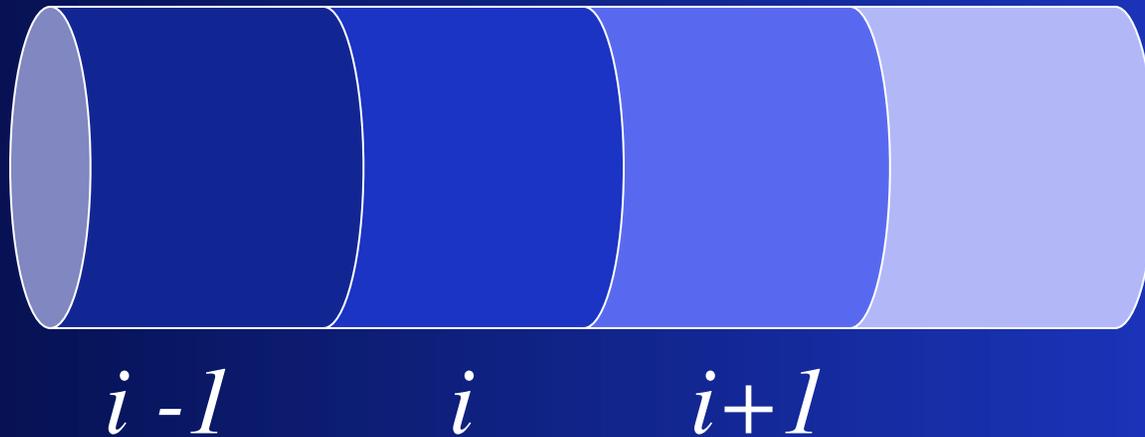
$$\frac{\partial c(x, t)}{\partial t} = \frac{D}{\Delta x^2} [c(x - \Delta x, t) - 2c(x, t) + c(x + \Delta x, t)]$$

Discrete Diffusion Equation



$$\frac{\partial c(x, t)}{\partial t} = \frac{D}{\Delta x^2} [c_{i-1} - 2c_i + c_{i+1}] + \sigma$$

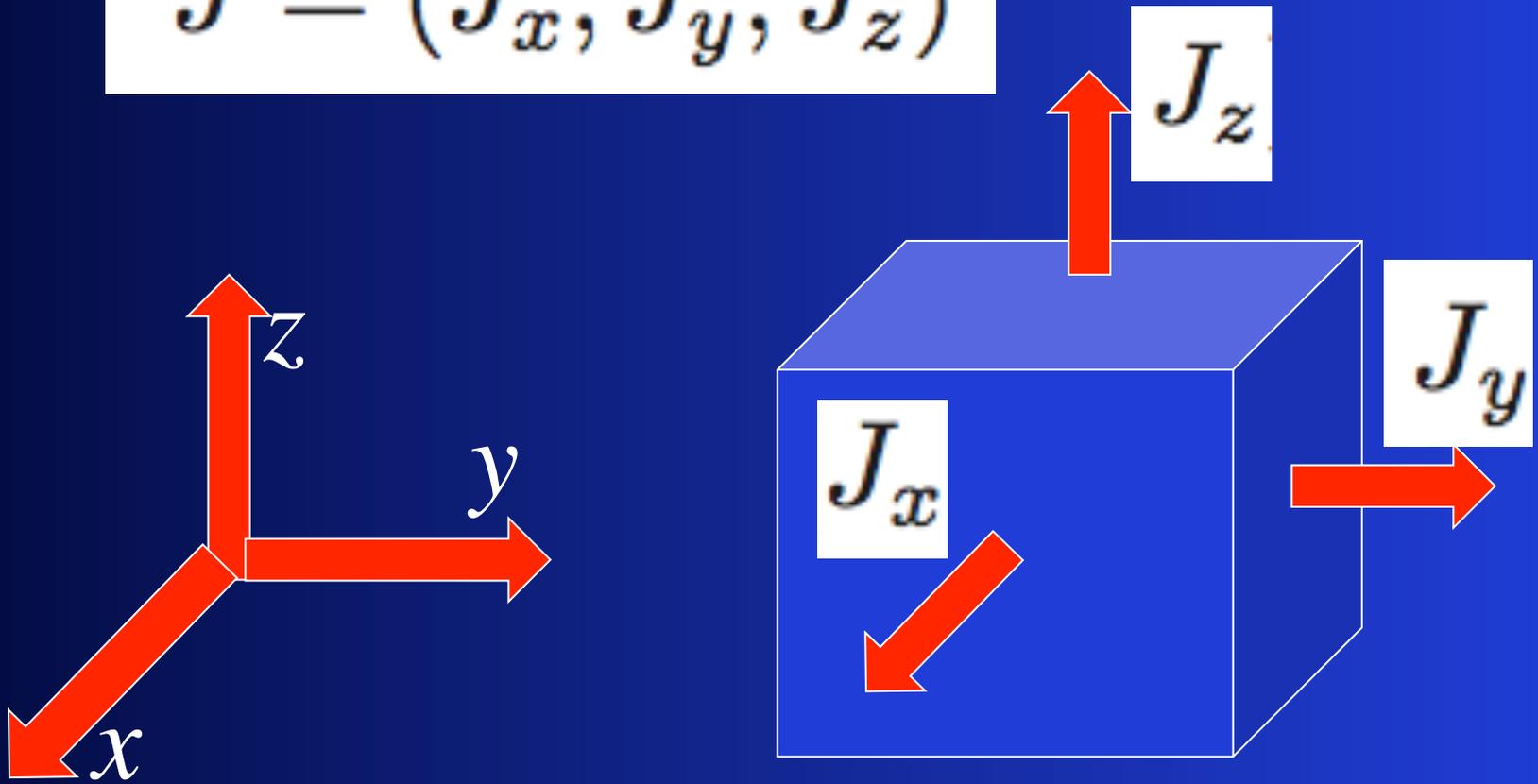
Discrete Diffusion Equation



$$\frac{\partial c(x, t)}{\partial t} = \frac{D}{\Delta x} \left(\left[\frac{c_{i+1} - c_i}{\Delta x} \right] - \left[\frac{c_i - c_{i-1}}{\Delta x} \right] \right)$$

Higher dimensions

$$\vec{J} = (J_x, J_y, J_z)$$



Higher dimensions

$$\vec{J} = (J_x, J_y, J_z)$$

$$\nabla = \left(\frac{\partial}{\partial x}, \frac{\partial}{\partial y}, \frac{\partial}{\partial z} \right)$$

$$\frac{\partial c}{\partial t} = -\nabla \cdot \vec{J}$$

Higher dimension: Diffusion

$$\vec{J} = -D\nabla c$$

$$\frac{\partial c}{\partial t} = -\nabla \cdot (-D\nabla c)$$

$$\frac{\partial c}{\partial t} = D\nabla^2 c$$

$$\frac{\partial c}{\partial t} = D\Delta c + \sigma$$

Higher dimension: transport

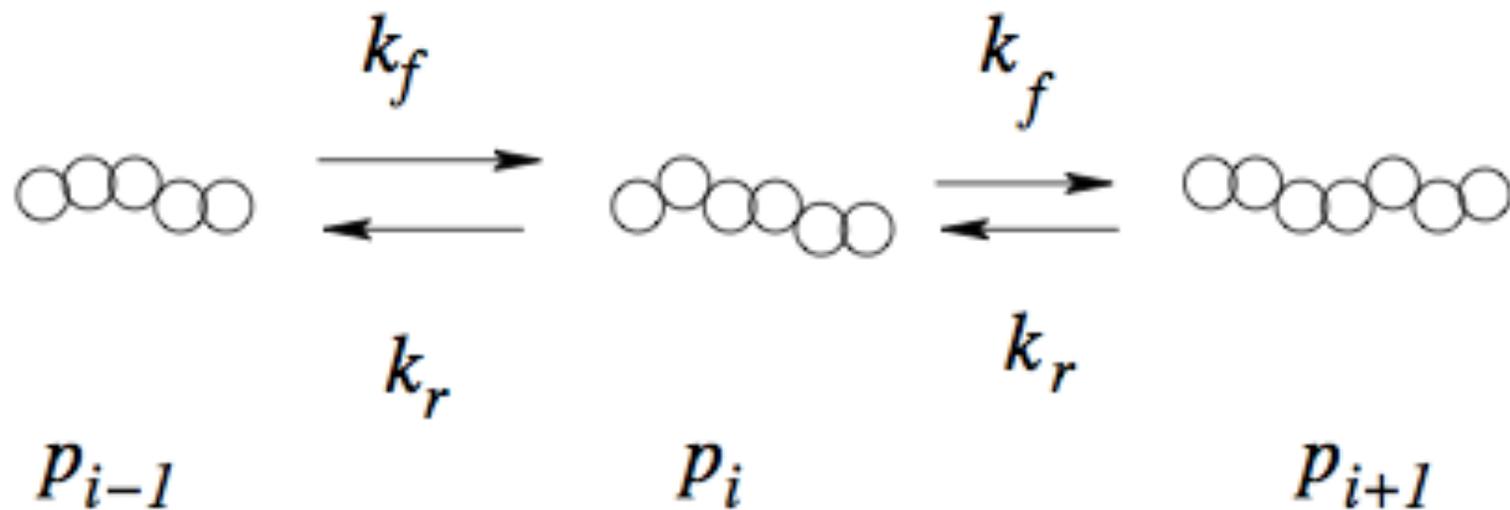
$$\vec{J} = \vec{v}c$$

$$\frac{\partial c}{\partial t} = -\nabla \cdot (\vec{v}c) + \sigma$$



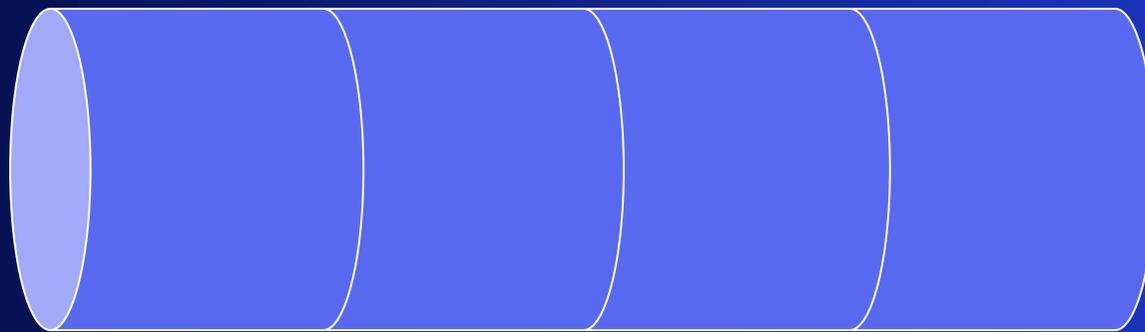
Polymer size distribution

Size classes



$$k_f = k^+ a, \quad k_r = k^-$$

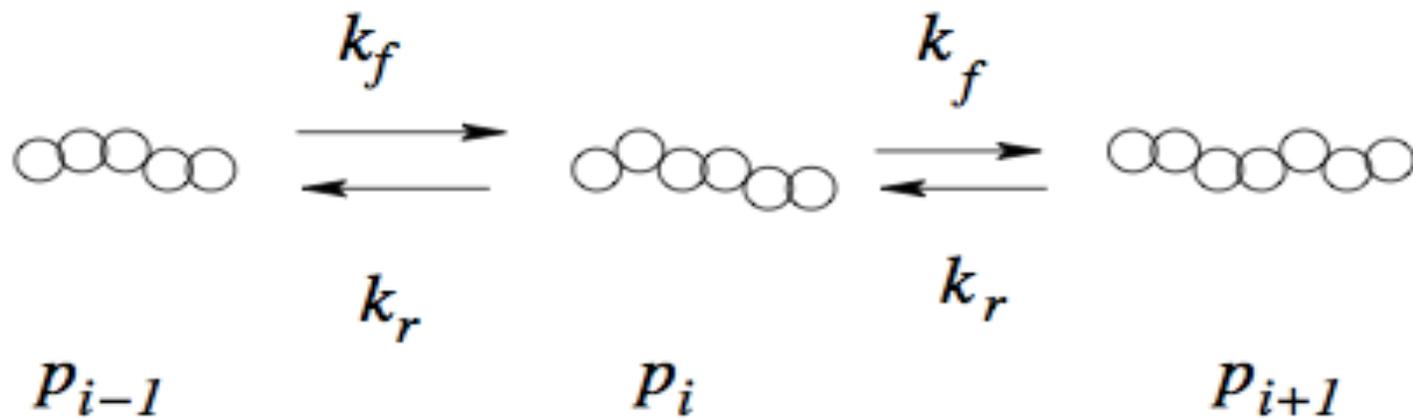
Discrete size classes



$i-1$

i

$i+1$



P_{i-1}

P_i

P_{i+1}

Balance equation

$$\frac{dp_i}{dt} = ck_f p_{i-1} - (ck_f + k_r) p_i + k_r p_{i+1}$$

